

SYNTHESIS AND ANALGESIC POTENTIAL OF SOME SUBSTITUTED PYRROLIDINES

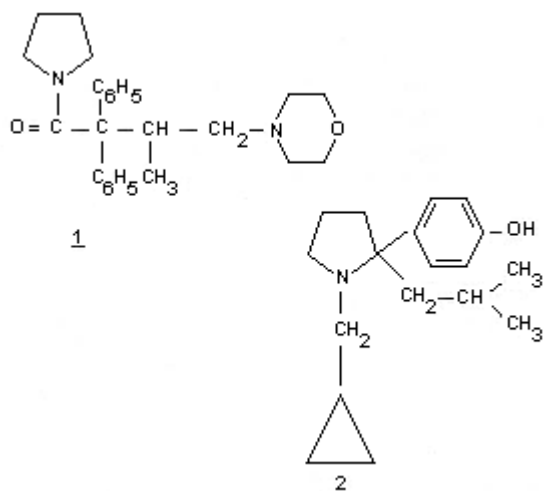
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SUMMARY: A series of 3-aryloxy-1-methyl-5-phenyl pyrrolidines (5a-k) was prepared and screened for their analgesic potency as well as their local anaesthetic activity.

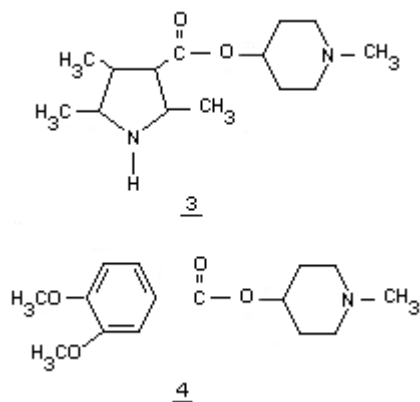
Key Words: Pyrrolidines, esters of 1-methyl-5-phenyl-3-pyrrolidinol analgesics, local anaesthetics.

INTRODUCTION

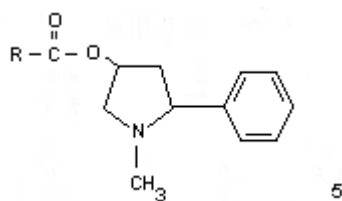
It has been reported that various substituted pyrrolidines display versatile pharmacological profiles, such as anticholinergics by Adamson *et al.* (1) and CNS stimulants by Seeger and Kottler (2), as antihistaminics by Weidmann *et al.* (3), Ondelli *et al.* (4), and antihypertensives by Rubin *et al.* (5). Also, among these profiles is the analgesic activity which has been disclosed by Keats *et al.* (6) in dextromoramide 1 and by Bowman *et al.* (7) in compound 2, which are more potent than morphine.



On the other hand, Waters (8-10) reported that simple aromatic and heteroaromatic esters of non-quaternary carbon-4 piperidinols 3 and 4 exhibit potent antinociceptive potential comparable to that of morphine without featuring physical dependence liability.

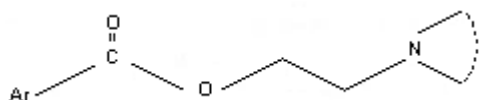


The present work describes the synthesis of some esters 5, of the novel 5-methyl-5-phenyl-3-pyrrolidinol 6, aiming to achieve compounds having antinociceptive properties.



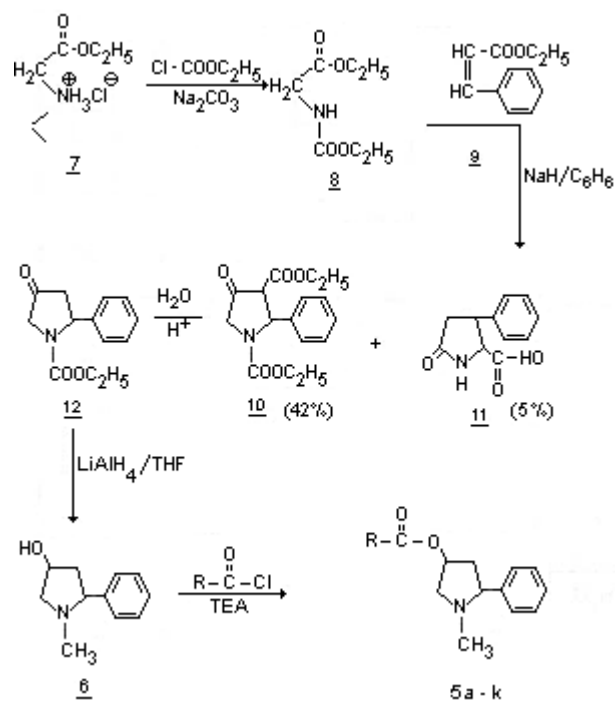
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Moreover, the prepared compounds exhibit the structural features of the local anaesthetics basic esters having the general skeleton:



Therefore, the biological investigation is extended to involve the local anaesthetics activity.

The synthetic strategy of the target compounds is illustrated in scheme 1 (vide infra).



Scheme 1

MATERIALS AND METHODS

Melting points were determined on Büchi 510 melting point apparatus and are uncorrected. Infrared spectra were recorded on Beckmann Infracord, model 4220. The microanalytical data were obtained from the National Research Centre, Dokki, Cairo. The analytical results deviated maximally $\pm 0.4\%$ from the theoretical values for C, H and N. The $^1\text{H-NMR}$ (CDCl_3 , TMS) were recorded on VARIAN (90 MHz ^1H). The mass spectrum was obtained on VARIAN CH-5 and CH-7 spectrometers. The mass spectrum was performed at the Institut für Pharmazeutische Chemie, Universität Munster, FRG. Purification of the products were performed by using column chromatography using neutral alumina activity I as an adsorbent and n-hexane / CH_2Cl_2 (1:1) as the eluent. Compounds **7** and **8** were prepared as previously described by Marver, Brenner *et al.* and Mair *et al.* (11-13) respectively.

Diethyl 4-oxo-2-phenyl-1, 3-pyrrolidine dicarboxylate **10**.

To a stirred suspension of 2.4 g (0.1 mol) of sodium hydride in 100 ml of dry benzene, 17.5 g (0.1 mol) of the diester **8** was added dropwisely. After the addition of one-quarter of **8**, the mixture was gently heated, to initiate the reaction. The rest of the diester **8** was added at such a rate to maintain gentle reflux throughout the addition. The reaction mixture was stirred overnight at room temperature, then 17.6 g (0.1 mol) of ethyl cinnamate **9** was added dropwisely and stirred for additional 30 min. The reaction mixture was refluxed for 2 h, then treated with 2 ml of absolute ethanol and refluxed for further 18 h. After cooling, the whole was poured over 300 ml of ice-water. The benzene layer was separated, and washed with water (2x50 ml). The combined aqueous layers were shaken with ether (2x50 ml), then acidified to pH 3-4 with 10% H_2SO_4 , extracted with ether (4x50 ml), dried (MgSO_4) and concentrated in vacuo where the cyclized product **10** (14-16) was crystallized from ether as white powder of 12.8 g (42%), mp 108-110°C.

IR (KBr, cm^{-1}): 3300 (OH), 1700, 1680 (C = O, esters) and 1640 (C = O of β -keto ester).

$^1\text{H-NMR}$ (CDCl_3 , TMS): $\delta(\text{ppm})=0.9-1.36$ (two t overlapping, 6H, two CH_3), 3.53-4.43 (broad m, 7H, two CH_2 and H in the 3-position), 5.53-5.71 (m, 1H, H in the 2-position), 7.23 (s, 5H, aromatic protons).

The aqueous acidic phase was concentrated to one-third of its volume, where the by product 3-phenyl-5-oxo-pyrrolidin-2-carboxylic acid **11** was precipitated as white crystals (5%), mp 184-6°C.

IR (KBr, cm^{-1}): 3200-3220 (NH, OH), 1730 (C = O, carboxylic acid) and 1660 (C = O, amidic carbonyl group).

Ethyl 4-Oxo-2-phenyl-1-pyrrolidine carboxylate **12**

A mixture of 30.5 g (0.1 mol of **10**, 90 ml water and 0.9 ml concentrated hydrochloric acid, was refluxed under stirring for 15 h. The resulting reaction mixture was saturated with NaCl and extracted with CH_2Cl_2 (3x10 ml), dried (MgSO_4) and evaporated in vacuo to give 23.3 g (100%) of **12** as reddish brown oil which was purified by column chromatography to give 22 g (95%) of **12** as viscous oil (15).

IR (liquid film, cm^{-1}): 1770 (C = O, five membered ring ketone) and 1700 (C = O ester).

$^1\text{H-NMR}$ (CDCl_3 , TMS): $\delta(\text{ppm})=1.16-1.3$ (t, 3H- $\text{NCOOCH}_2\text{CH}_3$) 2.56-3.4 (broad m, 2H, CH_2 in the 3-position), 4.0-4.4 (broad m, 4H, CH_2 in the 5-position and $-\text{NCOOCH}_2\text{CH}_3$), 5.56-5.73 (m, 1H, CH in the 2-position), 7.36-7.56 (m, 5H, aromatic protons).

The semicarbazone of **12**, mp 196-8°C (17).

1-Methyl-5-phenyl-3-pyrrolidinol **6**

A solution of 23.3 g (0.1 mol) of **6** in 60 ml of dry THF was

added dropwise to a stirred cold suspension of 7.6 g (0.2 mol) of LiAlH_4 in 120 ml of dry THF. The mixture was refluxed for 24 h, then cooled and the complex formed was decomposed with a saturated solution of Na_2SO_4 . After filtration, the solvent was evaporated in vacuo to give a viscous oil, which was acidified under cooling with 5N HCl and washed with ether (2x25 ml). The aqueous acidic layer was basified under cooling with 10 % NaOH, extracted with CH_2Cl_2 (3x30 ml), dried (MgSO_4) and evaporated in vacuo to give 15 g (85%) of **6** as dark brown oil, which was purified by column chromatography to give 14.5 g (82 %) as golden viscous oil. Analy ($\text{C}_{11}\text{H}_{15}\text{NO}$).

	C	H	N
Calc.	74.54	8.52	7.90
Found	74.80	8.70	7.80

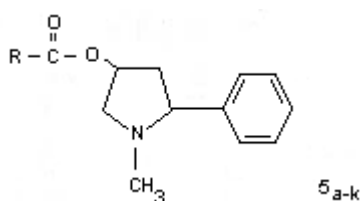
IR (liquid film, cm^{-1}): 3400 (OH, alcohol) and absence of carbonyl ester group.

$^1\text{H-NMR}$ (CDCl_3 , TMS) = δ (ppm) = 0.89-1.86 (broad m, 2H, CH_2 in the 4-position), 2.09 (s, 3H, CH_3 in the 1-position) 2.29-2.76 (m, 2H, CH_2 in the 2-position), 2.86-3.19 (M, 2H, CH and OH in the 3-position, OH exchangeable with D_2O), 4.13-4.33 (m, 1H, CH in the 5-position) 7.29 (s, 5H, aromatic protons).

General procedure for the preparation of 3-aroxyloxy-1-methyl-5-phenyl pyrrolidines (5a-k, c.f. Table 1).

To a solution of 1.77 g (0.01 mol) of **6** and 3g (0.03 mol) of triethylamine in 60 ml dry benzene, was added (0.015 mol) of the appropriate acid chloride. The mixture was refluxed and stirred overnight. The precipitated triethylamine hydrochloride was filtered off and washed with benzene (25 ml). The combined filterates

Table 1



No. of Comp	R	mp, °C	Yield %	Formula (Mol. Wt)	Analysis %			
					Calc.	C	H	N
5a	C_6H_5	73-5b	50	$\text{C}_{18}\text{H}_{19}\text{NO}_2$ (281.36)	Calc.	76.84	6.80	4.97
					Found	76.65	7.06	5.21
5b	4-Cl- C_6H_4	107-9b	40	$\text{C}_{18}\text{H}_{18}\text{ClNO}_2$ (315.78)	Calc.	68.46	5.74	4.43
					Found	67.85	5.97	4.13
5c	4- CH_3O - C_6H_4	106-8c	69	$\text{C}_{19}\text{H}_{21}\text{NO}_3\text{HCl} \cdot 1.3\text{H}_2\text{O}$ (371.26)	Calc.	61.46	6.67	3.77
					Found	61.52	6.44	3.83
5d	4- NO_2 - C_6H_4	58-7b	76	$\text{C}_{18}\text{H}_{18}\text{N}_2\text{O}_4$ (326.35)	Calc.	66.24	5.56	8.58
					Found	66.10	5.23	8.12
5e	4- NH_2 - C_6H_4	97-9b	91	$\text{C}_{18}\text{H}_{20}\text{N}_2\text{O}_2$ (296.36)	Calc.	72.94	6.80	9.45
					Found	73.19	6.87	9.27
5f	2,4-(Cl) $_2$ - C_6H_3	58-60b	60	$\text{C}_{18}\text{H}_{17}\text{Cl}_2\text{NO}_2$ (350.24)	Calc.	61.72	4.89	3.99
					Found	61.70	4.20	3.93
5g	3,4,5-(CH_3O) $_3$ - C_6H_2	127-9c	65	$\text{C}_{21}\text{H}_{25}\text{NO}_5 \cdot \text{HCl} \cdot 2\text{H}_2\text{O}$ (443.93)	Calc.	56.81	6.81	3.15
					Found	56.86	6.97	3.05
5h	C_6H_5 - CH_2	95-7b	64	$\text{C}_{19}\text{H}_{21}\text{NO}_2$ (295.38)	Calc.	77.25	7.16	4.74
					Found	77.30	6.90	4.60
5i		Oild	68	$\text{C}_{24}\text{H}_{29}\text{NO}_2$ (363.49)	Calc.	79.30	8.04	3.85
					Found	79.60	7.80	3.80
5j		120-2b	70	$\text{C}_{25}\text{H}_{25}\text{NO}_2$ (371.47)	Calc.	80.83	6.78	3.77
					Found	80.60	7.05	4.01
5k		Oild	52	$\text{C}_{24}\text{H}_{31}\text{NO}_2$ (365.51)	Calc.	78.86	8.54	3.83
					Found	79.20	8.35	3.62

a) IR (KBr or liquid film, cm^{-1}) of 5a-k: 1730-1700 (C=O, ester)


b) Free base: which were crystallized from ether.

c) Hydrochloride: which were crystallized from 2-Propanol: ether.

d) Were separated as viscous oil and purified by column chromatography.

were evaporated in vacuo to give an oily product, which was treated with 5N HCl and the undissolved substance was extracted with ether (2 x 25 ml). The aqueous acidic layer was rendered alkaline with 10% NaOH extracted with CH₂Cl₂ (3 x 15 ml), dried (MgSO₄) and evaporated in vacuo (c.f. Table 1).

Compound **5e** was obtained by catalytic hydrogenation of an ethanolic solution of **5d**, using PtO₂ as catalyst.

5b: MS (m/z): 316 (M⁺C₁₈H₁₈NO₂, 1%) base peak (M⁺-157  100%). ¹H-NMR (CDCl₃, TMS): δ(ppm) of **5h**: 1.02-1.3 (m, 2H, CH₂ in the 4-position) 2.2 (s, 3H, CH₃-N), 2.47-2.8 (m, 2H, CH₂ in the 2-position), 3.0-3.43 (m, 1H, CH in the 3-position), 3.73 (s, 2H, CH₂-C₆H₅), 5.16-5.4 (m, 1H, CH in the 5-position), 7.4-7.56 (m, 10H, aromatic protons).

Pharmacology

Analgesic activity:

The test compounds **5a-k** were evaluated for their analgesic activity using the hot-plate technique as described by Janssen *et al.* (18). Morphine HCl was used as reference. The dose producing analgesic activity in 50% of mice (ED₅₀ and 95% confidence limits) was calculated by the graphical method of Litchfield and Wilcoxon (19).

Local anaesthetic activity:

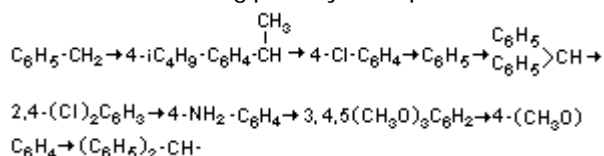
The procedure of Jones and Weaver (20) was adopted, using the cornea of the male albino mice and procaine hydrochloride as a reference six mice were used to locate each point on the dose response curve which was calculated by the method of Litchfield and Wilcoxon (19).

RESULTS AND DISCUSSION

Analgesic activity

Compounds **5a-k** were evaluated for their analgesic activity (18) (Table 2). The most active compounds in this series are **5h** and **5k**. Both possess an analgesic effect about 6.4 % that of morphine hydrochloride.

With respect to the structure-activity relationship (SAR), the substitution in the aromatic moiety of **5** revealed the following potency order preference:



Local anaesthetic activity

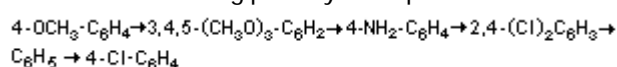
Using albino mice for local anaesthetic evaluation (20) (Table 2) it was found that compounds **5c** and **5g** exhibited stronger local anaesthetic potency than the reference drug procaine hydrochloride. Compounds **5c** and **5g** pos-

Table 2

Compound No.	Analgesic activity ED ₅₀ mg/kg (95% confidence limit)	Local anaesthetic activity ED ₅₀ mg/kg (95% confidence limit)
5a	48.8 (28.10-76.00)	58.4 (42.62-80.00)
5b	40.0 (28.70-55.60)	64.00 (52.89-77.44)
5c	63.6 (53.44-75.68)	7.6 (6.44-8.97)
5e	58.4 (48.26-70.66)	48.4 (28.30-82.76)
5f	57.2 (42.06-77.79)	53.6 (43.93-65.39)
5g	62.8 (50.24-79.00)	16.8 (8.40-34.40)
5h	34.5 (30.63-38.84)	-
5i	47.2 (32.55-68.44)	-
5j	64.4 (59.08-70.19)	-
5k	35.0 (22.70-54.00)	-
Reference	Morphine HCl	Procaine HCl
drug	2.23 (1.65-3.01)	42.5 (24.15-74.70)

sessed local anaesthetic potency 5.6 and 2.5 times that of the reference drug, respectively.

With respect to the structure-activity relationship (SAR) the substitution in the aromatic moiety of **5** revealed the following potency order preference:



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