Biochemistry

THE ROLE OF HYDROXAMIC ACIDS IN BIOCHEMICAL PROCESSES

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SUMMARY: Hydroxamic acids, a group of naturally occurring and synthetic weak organic acids of general formula RC(=O)N(R')OH, are widespread in the tissues of plants, in metabolites of bacteria and fungi, including complex compounds. Hydroxamic acids and their derivatives fulfill a variety of important roles in biology and medicine; here we provide a comprehensive brief review of the most basic medicinal chemistry and pharmacology of hydroxamate molecules.

Key Words: Hydroxamic acids, biological activities, biochemical processes.

INTRODUCTION

Since their discovery by Wahlroos and Virtanen (1) in 1959, and over the past decades, the chemistry and biochemistry of hydroxamic acids and their derivatives have attracted considerable attention, due to their pharmacological, toxicological and pathological properties. Hydroxamic acids generally have low toxicities and have a wide spectrum of activities in all types of biological systems, as such they act variously as growth factors, food additives, tumor inhibitors, antimicrobial agents, antituberculous, antileukemic agents, key pharmacophore in many important chemotherapeutic agents, pigments and cell-division factors. Several of them have been advanced into human clinical trials as pharmaceutical drugs, for the treatment of several diseases. The following is a brief concerted attempt to describe the above roles of hydroxamate molecules in a variety of circumstances that are used widely in biology and medicine.

INHIBITION EFFECT AND ANTICANCER ACTIVITY OF HYDROXAMIC ACIDS

The design and synthesis of ligands for biomedical applications in fields such as anticancer applications has become of great importance. One of these important ligands is hydroxamate molecules. Hydroxamic acids have been found to react with both proteins and nucleic acids (2). The reactivity of hydroxamic acids towards sulfhydryl groups of proteins has been suggested to be the reason for their inhibitory effect on various enzymes. The protease papain, for instance, with a single free cysteine residue located at the active site was irreversibly inhibited by DIMBOA (2,4-dihydroxy-7-methoxy-1,4-benzoxazin-3one). Friebe and co-workers (3) showed an inhibitory effect of DIBOA (2,4-dihydroxy-1,4-benzoxazin-3-one) and DIMBOA on plasma membrane H+-ATPase from roots A. sativa and Avena fatua. This inhibition may also be due to the reactivity of hydroxamic acids towards sulfhydryl groups since at least one exposed cysteine residue at the active site is of importance for maintenance of enzyme conformation (3). In addition, DIMBOA was shown to have an inhibitory effect on the electron trans-

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port and thus adenosine-tri-phosphate (ATP) production in isolated mitochondria and chloroplasts of maize (4). Both DIBOA and DIMBOA have been shown to be mutagenic in a test with *Salmonella typhimurium* (2). The reactivity of hydroxamic acids offers an explanation to the various biological effects observed.

The matrix metalloproteinases (MMP) are a family of zinc-dependent enzymes that are required for extra cellular matrix degradation and tissue remodeling. The ability of the hydroxamic acid functionality to form abidentate chelate with the zinc and nickel atoms in the enzyme's active site is considered to be an important functional feature metalloenzyme inhibition, namely as inhibitors of metalloproteinase (5), matric metalloproteinase (6, 7), they are also potent and specific inhibitors of urease activity (8, 9), thermolysin (10, 86), elastase (11), peroxidases (12), amino peptidases (13, 87).

It is well known from the literature (14–20, 88) that the unsubstituted aliphatic hydroxamic acids (such as acetohydroxamic acid) are well established as effective inhibitors of plant (14,15, 88), and bacterial urease *in vitro* (15,16) and have been shown to effectively inhibit ureolytic activity and/or to lower blood ammonia levels in mice (17), rats (15), sheep (18), cows (19), dogs and men (20). The potential application of these compounds in the treatment of hepatic coma and in the improvement of nitrogen utilization by ruminant animals has led to the present series of their physiologic disposition in the animal body. The four lower aliphatic hydroxamates were studied by chromatographic and spectrophotometric methods in unlabeled form (21).

Recently, significant advances have been made toward understanding the inhibition phenomena basis of hydroxamic acids. A number of hydroxamic acid analogues have been shown to inhibit DNA (dinucleic acid) synthesis by inactivating the enzyme ribonucleotide reductase (RNR) (22–25). This metalloenzyme catalyzes the conversion of (ribo) nucleotides to deoxy (ribo) nucleotides and is therefore a potential target for the development of anticancer agents (26–28). Hydroxamic acid moiety, R-CONHOH, is found to be the essential pharmacophore in the hydroxyurea, a clinically useful inhibitor of ribonucleotide reductase (29). A variety of nucleoside analogues are also active as inhibitors of ribonucleotide reductase (30, 32), following the inhibition mechanism similar to that proposed for hydroxamates (33-35). Farr *et al.* (37) designs and synthesizes a nucleoside analogue incorporating hydroxamate moiety. Their compounds inhibited RNR activity, but were 10-fold less potent than hydroxyurea. Moreover, recent reports (37–39) show that hydroxamate compounds increase the potency of nucleosides against HIV-1 (human-immunod-eficiency-virus-1) give an additional importance to derivatives combining the structural features of the above compounds.

Quite recently, there have been articles concerning the biochemistry of anticancer activity of naturally occurring and synthetic class of organic compounds containing the hydroxamic acid functional group (-CONHOH). Hydroxyurea containing that group, is a well known anticancer drug (40-43). It inhibits the DNA synthesis by imparing the activity of enzyme ribonucleotide reductase (42-45). Though it is clinically used as anticancer agent, it perturbs the hematological parameters and depresses the bone marrow (46). Subsequently anticancer properties of some aliphatic and aromatic hydroxamic acids (such as acetohydroxamic acid, benzohydroxamic acid and salicylhydroxamic acid) have been studied (47, 48). Recently it has been reported that cholorohydroxamic acid possesses antitumor properties and inhibits the growth of Ehrlich ascites carcinoma (EAC) cells by imparing DNA and protein synthesis without altering the hematological parameters (49). No such studies have yet been done with acetohydroxamic acid, benzohydroxamic acid and salicylhydroxamic acid. The comparative study of antineoplastic activities and host toxic effects of hydroxamic acids (cholorohydroxamic acid, acetohydroxamic acid, benzohydroxamic acid, salicylhydroxamic acid and hydroxyurea) has been recently reported (50). In addition peritoneal macrophages and lipid peroxidation in normal mice after treatment with these hydroxamic acids have been presented (51). Transplantability of hydroxamic acid treated EAC cells has also been observed. The results show that cholorohydroxamic acid can be considered as the most effective antitumor agent amongst the hydroxamic acids studied and are comparable with hydroxyurea regarding cell growth inhibition and survival time of tumor bearing mice. However, it is necessary that the antitumor activity of cholorohydroxamic acid should be carried out against different tumor cell lines which may bring promising results in cancer chemotherapy.

Recently a new oral chelator, salicylhydroxamic acid was developed and found to have promising advantages in the clinical treatment of thalassaemia major (52), as a trypanocidal drug (53) and used as inhibitors of viral growth (54), selective inhibition of deoxyribonucleic acid synthesis (55), also salicylhydroxamic acid inhibits delta 6 desaturation in the microalgo porphyridium cruentum (56) and also, the selective inhibition of catechol oxidases by salicylhydroxamic acid was reported (57). In addition to this, halogen substituted salicylhydroxamic acids used in lowering of rabbits (58).

In recent studies, it was reported that, benzohydroxamic acid had significant antitumor activity (59), substituted benzohydroxamic acid has been prepared to enhance the effect of benzohydroxamic acid (60, 90, 91) and its complexes with copper metal ions (Cu-benzohydroxamic acid) and used as a potential antitumor drug (61).

Because of the wide spread physiological importance of trihydroxamic acids, it is highly desirable to investigate several natural and synthetic trihydroxamic acids that will be selective for therapy of certain diseases. Desferrioxamine, a natural trihydroxamic acid, is a chelating of iron, aluminum and other metals, is used therapeutically for the treatment of iron-overloaded-patients (62, 92) to remove excess iron in patients suffering from iron overload as a consequence of the treatment of Cooley's anemia, or acute iron poisoning, particularly in patients with AIDS, but the lack of oral activity and its short biological half-life limits its use. Otherwise, iron chelation by desferrioxamine, and other chelators, protects against the cytotoxic and reactivating effects of hydrogen peroxide (63), and thus decreases NF-kB activation of HIV-1 transcription. Also, desferrioxamine B is used in human medicine removal of excess aluminum from the human body in those patients who must undergo permanent hemodialysis (64, 93).

More recently oxazole and oxadiazole hydroxamic acids are claimed by Pfizer as inhibitors of procollagen C-

proteinase (PCP or pCP). Fundamental work on this enzyme has been carried out at Thomas Jefferson University, who described recombinant pCP and its use in treatment of fibrotic disorders. Until now, virtually the only indication that pCP inhibitors were targets for systematic drug discovery were from Roche, whose preferred candidates are also heterocyclyl hydroxamic acids, for example thiazoles.

Allelopathy, the chemical interaction of plants within the same species or between plants of different species, is important in the plant competition for water, nutrients and light. Hydroxamic acids have been ascribed a role in this interaction in many reports. DIBOA and BOA (2benzoxazolinone) were shown to have an inhibitory effect on root growth of cress (*Lepidium sativum*) and barnyardn grass (*Echinochloa crusgalli*) (65). DIMBOA and MBOA (6-methoxy-2-benzoxazolinone) from *Triticum durum* were shown to have a growth inhibiting effect on roots of the weed *A. fatua* (66). In addition, MBOA was shown to inhibit seed germination on *A. fatua*. Rye root exudates containing DIBOA inhibited root growth of *A. fatua* whereas wheat root exudates without detectable amounts of hydroxamic acid did not (67).

A possible explanation to the growth inhibitory effect of hydroxamic acids on other plant species might be due to their ability to modify auxin action. Venis and Watson (68) reported that methylated benzoxazolinones are able to inhibit the binding of auxin to membrane receptors. MBOA was shown to have an inhibiting effect on auxin-induced bending (at concentrations of 0.6 mM or higher) and elongation (at concentrations of 0.6 mM) of oat coleoptiles (69). In contrast, DIMBOA had a supporting effect on auxininduced elongation of maize coleoptiles at a concentration of 20 mM (70). The differences between the results of these experiments might be explained by different sensitivities of the plant species tested. Maize already contains DIMBOA and a UDP-glucose: Hx(hydroxamic acids)-glucosyltransferase (71, 72), to detoxify DIMBOA, that might interfere with the experiment. For instance, it has been shown that A. thaliana transformed with the genes for UDP-glucose: hydroxamic acid - glucosyltransferases isolated from maize are less sensitive to the allelopathic substances DIBOA and DIMBOA in growth assays than wt A. thaliana that does not contain glucosyltransferases acting on these substances (72). Another possible explanation is that in maize only MBOA and not DIMBOA is a potent inactivator of auxin-induced shoot elongation (73). Maize bxbx mutant, deficient in DIMBOA synthesis grow normally, although extremely susceptible to pathogen attack (74). It has been shown, though, that the bxbx mutant still contains trace amounts of DIMBOA, which might be enough to maintain a normal growth (75). However DIMBOA, when exuded from Hx/HxGlc (hydroxamic acids/hydroxamic acid glucoside(s)) containing plants, may be more important in plant competition with other plants for water, light and nutrients by inhibiting the growth of neighboring plants than in growth regulation of the plant itself.

RESISTANCE ACTIVITY OF HYDROXAMIC ACIDS TOWARDS INSECTS

Hydroxamic acids have been shown to have a negative impact on the survival and reproduction of aphids. Argando-a et al. (76) found inverse correlations between hydroxamic acid content in different varieties of rye and wheat and the growth rate of the aphid Metopolophium dirhodum. When the plants grow older and the hydroxamic acid levels become decreased, the growth rate of the aphid populations are increased. In the same paper it was also reported that aphids fed with artificial diets containing DIMBOA or MBOA had lower survival rate than aphids fed on artificial diets lacking DIMBOA or MBOA. Correlations have also been shown in maize between high hydroxamic acid levels and resistance to the European corn borer Ostrinia nubilalis (77, 78). In addition to the reports based on correlations and work with insects on artificial diets, hydroxamic acids have been shown to be induced by infestation with insects. Hydroxamic acid levels in the wounded tissue of maize stems and leaves increased upon infestation with larvae of the corn borer Sesamia nonagrioides (79). In wheat, several cultivars including one T. durum cultivar showed increased levels of DIMBOA in leaf tissue upon infestation with the aphids M. dirhodum and Rhopalosiphum padi (2, 80). Induced changes in hydroxamic acids levels were however, later shown in wild wheat (Triticum uniaristatum) to be due to translocation rather than enhanced local synthesis of Hx (81).

ANTIFUNGAL ACTIVITY OF HYDROXAMIC ACIDS

As early as 1959, Wahlroos and Virtanen (1) reported about the anti-fungal effect of hydroxamic acids and their breakdown products on snow mold (Fusarium nivale). Fungi grown on medium containing DIMBOA, MBOA or BOA showed smaller colony diameter than fungi grown on an identical medium but lacking these compounds. Later reports indicated inverse correlations between infection ratings of Northern corn leaf blight-producing fungus (Helminthosporium turcicum) and plant DIMBOA levels, as well as inhibition of H. turcicum spore germination by DIMBOA (82, 83). These and other inhibitory effects of Hx on fungal growth are summarized by Niemeyer et al. (2). More recent studies of the fungal pathogen Gaeumannomyces graminis that causes the disease take-all in wheat and barley showed that DIBOA was a more potent fungal growth inhibitor than DIMBOA which is in correlation with the resistance of rye to take-all (84).

ANTIBACTERIAL ACTIVITY OF HYDROXAMIC ACIDS

Bacterial stalk rot of maize is caused by a certain strain of *Erwinia chrysanthemi*. Maize is, however, resistant to rot caused by other isolates of *Erwinia chrysanthemi* and other soft rotting *Erwinia* species. It has been shown that DIMBOA inhibits the growth of several soft rotting *Erwinia* species at concentrations of 0.2-0.3 mM and that strains non-pathogenic to maize were more sensitive to DIMBOA than pathogenic strains. DIMBOA was therefore proposed to be involved in the resistance towards *Erwinia* (85).

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